## WHAT IS CLAIMED:

 $1. \quad \text{A platinum group metal-free (PGM-free)} \\ \\ \text{regenerable catalyst composition for entrapping $SO_x$, the } \\ \\ \text{composition comprising a component having formula I;} \\ \\$ 

Cu/(A oxide) I

wherein A oxide is selected from the group consisting of  $SiO_2$ ,  $Zr-SiO_2$ ,  $Al_2O_3$ ,  $TiO_2-Al_2O_3$ ,  $ZrO_2$ ,  $In_2O_3$ , and mixtures thereof.

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- 2. The platinum group metal-free regenerable catalyst of claim 1 wherein the Cu loading is from about 10 mol% to about 60 mol%.
- 15 3. The platinum group metal-free regenerable catalyst of claim 1 wherein the Cu loading is about 25 mol%.
- 4. The platinum group metal-free regenerable catalyst composition of claim 1 wherein the A oxide is selected from the group consisting of SiO<sub>2</sub>, Zr-SiO<sub>2</sub>, TiO<sub>2</sub>-Al<sub>2</sub>O<sub>3</sub>, ZrO<sub>2</sub>, In<sub>2</sub>O<sub>3</sub>, and mixtures thereof.
  - 5. The platinum group metal-free regenerable catalyst composition of claim 1 wherein the A oxide is selected from the group consisting of SiO<sub>2</sub>, Zr-SiO<sub>2</sub>, and mixtures thereof.
- 6. The platinum group metal-free regenerable catalyst composition of claim 1 wherein the A oxide comprises 30 SiO<sub>2</sub>.
  - 7. The platinum group metal-free regenerable catalyst composition of claim 1 wherein the A oxide comprises  $Zr-SiO_2$ .

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- 8. A method of adsorbing  $SO_x$  as metal sulfate in a temperature range of 200°C to 500°C under lean fuel conditions using the catalyst composition of claim 1.
- 9. A method of desorbing metal sulfates at a temperature range of 250°C to 450°C under rich fuel conditions using the catalyst composition of claim 1.
- 10. A platinum group metal free catalyst composition for entrapping  $SO_x$ , the catalyst composition comprising an oxide selected from the group consisting of praseodymia, zirconia-praseodymia and mixed manganese-yttria and mixtures thereof.
- 15 11. The catalyst composition of claim 10 comprising praseodymia.

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- 12. The catalyst composition of claim 10 comprising zirconia-praseodymia.
- 13. The catalyst composition of claim 10 comprising mixed manganese-yttria.
- 14. A method of adsorbing  $SO_x$  as metal sulfate in 25 a temperature range of 200°C to 500°C under lean fuel conditions using the catalyst composition of claim 10.
- 15. A method of desorbing metal sulfates at a temperature range of 250°C to 450°C under rich fuel conditions using the catalyst composition of claim 10.
  - 16. A vehicle exhaust system comprising: a nitrogen oxide trap; and
- a  $\mathrm{SO}_x$  adsorbing component located upstream of the nitrogen trap in the vehicle exhaust system, the  $\mathrm{SO}_x$

adsorbing material comprising a catalyst selected from the group consisting of:

a) a platinum group metal-free (PGM-free) regenerable catalyst composition for entrapping  $SO_{\rm x}$  comprising a component having formula I;

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Cu/(A oxide) I

wherein A oxide is selected from the group consisting of  $SiO_2$ ,  $Zr-SiO_2$ ,  $Al_2O_3$ ,  $TiO_2-Al_2O_3$ ,  $ZrO_2$ ,  $In_2O_3$ , and mixtures thereof; or

- b) a platinum group metal free regenerable catalyst composition for entrapping  $SO_x$  comprising an oxide selected from the group consisting of praseodymia, zirconia-praseodymia and mixed manganese-yttria and mixtures thereof.
- 17. The vehicle exhaust system of claim 16 wherein the  $SO_x$  adsorbing material comprises a platinum group metal-free (PGM-free) regenerable catalyst composition for entrapping  $SO_x$  comprising a component having formula I;

Cu/(A oxide) I

- wherein A oxide is selected from the group consisting of  $SiO_2$ ,  $Zr-SiO_2$ ,  $Al_2O_3$ ,  $TiO_2-Al_2O_3$ ,  $ZrO_2$ ,  $In_2O_3$ , and mixtures thereof;
- 18. The vehicle exhaust system of claim 17 wherein the A oxide is selected from the group consisting of  $SiO_2$ ,  $Zr-SiO_2$ ,  $TiO_2-Al_2O_3$ ,  $ZrO_2$ ,  $In_2O_3$ , and mixtures thereof.
- 19. The vehicle exhaust system of claim 17 wherein the A oxide is selected from the group consisting of  $SiO_2$ , 30  $Zr-SiO_2$ , and mixtures thereof.
  - 20. The vehicle exhaust system of claim 16 wherein the  $SO_x$  adsorbing material comprises an oxide selected from the group consisting of praseodymia, zirconia-praseodymia and mixed manganese-yttria, and mixtures thereof;

- \$21.\$ The vehicle exhaust system of claim 16 wherein the  $SO_{x}$  adsorbing component is a diesel oxidation catalyst.
- 5 22. The vehicle exhaust system of claim 16 wherein the  $SO_{\kappa}$  adsorbing component is a catalyzed soot filter.